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> JP63013282A2: NONAQUEOUS ELECTROLYTE SECONDARY BATTERY

Electrode useful for non-aq. secondary batteries - consists of graphite-like

carbon deposited from gas phase [Derwent Record]

JP Japan

A (See also: JP06003745B4)

TAJIMA YOSHIMITSU;

MORI MOTOO: TANAKA HIDEAKI;

SHARP CORP **Assianee:

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1988-01-20 / 1986-07-02 Published / Filed:

> JP1986000156966 **P**Application

Number:

Advanced: C01B 31/02; C23C 16/26; H01M 4/58; H01M 4/96; H01M 10/40;

Core: C01B 31/00; H01M 10/36; more... IPC-7: H01M 4/60; H01M 10/40;

1986-07-02 JP1986000156966

PURPOSE: To increase charge-discharge cycle performance, to

limit discharge capacity by that of a negative electrode, and to widen the range of usable materials by making the chargedischarge capacity of a positive electrode larger than that of a negative electrode in a nonaqueous electrolyte secondary battery comprising a positive electrode, a nonaqueous electrolyte, and a negative electrode using a specific carbon as active material.

CONSTITUTION: As a negative electrode of a secondary battery

using nonaqueous organic solvent, a carbon body mainly comprising carbon which is formed from a hydrocarbon compound

by vapor phase build-up method by low temperature heat decomposition at 1500°C or lower and has planar net-like sixmembered ring atructure (graphite structure) having disorder layer structure and selective orientation structure is used, and the capacity of the negative electrode is made smaller than that of a positive electrode. A main planar spacing of the carbon is

0.337~0.355nm. The ratio of Raman strength 1360cm-1, to Raman

strength of 1580cm-1 in Raman structure is 0.4~1.0. V2O5, Cr2O3,

chalcogen compound, and composite or mixture of these

compounds are used in the positive electrode.

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PDF	Patent	Pub.Date	Inventor	Assignee	Title
怒	US6706447	2004-03-16	Gao; Yuan	i MO Corporation,	<u>Lithium metal dispersion in</u> secondary battery anodes

POther Abstract Info:

CHEMABS 108(10)078670M DERABS C87-272820







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